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## D1.5 Chevrel phases

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### D1.5.1 Introduction

Chevrel phase superconductors are a class of materials that have generated enormous interest in the superconductivity community. There are more than 100 different Chevrel phase compounds that exhibit a wide range of properties of both fundamental and technological interest. They have the chemical formula  $M_xMo_6X_8$  where  $x$  can range from 1 to 4, M can be one of more than 25 different elements and X is usually one of the chalcogenides (S, Se or Te). The structure can incorporate elements of different size, concentration and oxidation state. Some of the most intensively studied Chevrel phase materials with superconducting critical temperatures above 2 K are listed in table D1.5.1 [1–5]. Figure D1.5.1 shows the critical temperature, magnetic ordering temperature and some structural properties of Chevrel phase materials with rare-earth ions [1, 2, 6–8].

Chevrel phase compounds were discovered in 1971 [9]. Interest has been focused on the  $PbMo_6S_8$  and  $SnMo_6S_8$  materials with high upper critical fields ( $B_{c2} \sim 40\text{--}60\text{ T}$ ) and  $T_c$  values of  $\sim 12\text{--}15\text{ K}$ . The  $B_{c2}$  values of these Chevrel phase compounds lie between those of the high temperature copper oxide superconductors (e.g.  $Bi_2Sr_2Ca_2Cu_3O_x$ ) and the intermetallic low temperature superconductors (e.g.  $Nb_3Sn$ ) which means the superconducting coherence length is sufficiently long to reduce the effects of granularity found in high temperature superconductors but sufficiently short for very high field applications. Powder-in-tube wires have been fabricated (cf figure D1.5.2 [10–12]) with high critical current densities in high magnetic fields [10, 13, 14] in single lengths up to 1 km long (cf Chapter B3.3.5). This opens the possibility that these materials will be used in the next generation of high field applications operating at magnetic fields significantly above 25 T [12].

In addition to technological interest in these materials, the fundamental interest includes studies of many compounds which include rare-earth elements that are found to exhibit the coexistence of superconductivity and long range magnetic order [15–18] and studies of whether Chevrel phase compounds belong to a new class of (non-BCS [19]) superconductors that include the cuprates and is characterized by relatively high critical temperature for such small  $n_s/m^*$  (carrier density/effective mass) [20].

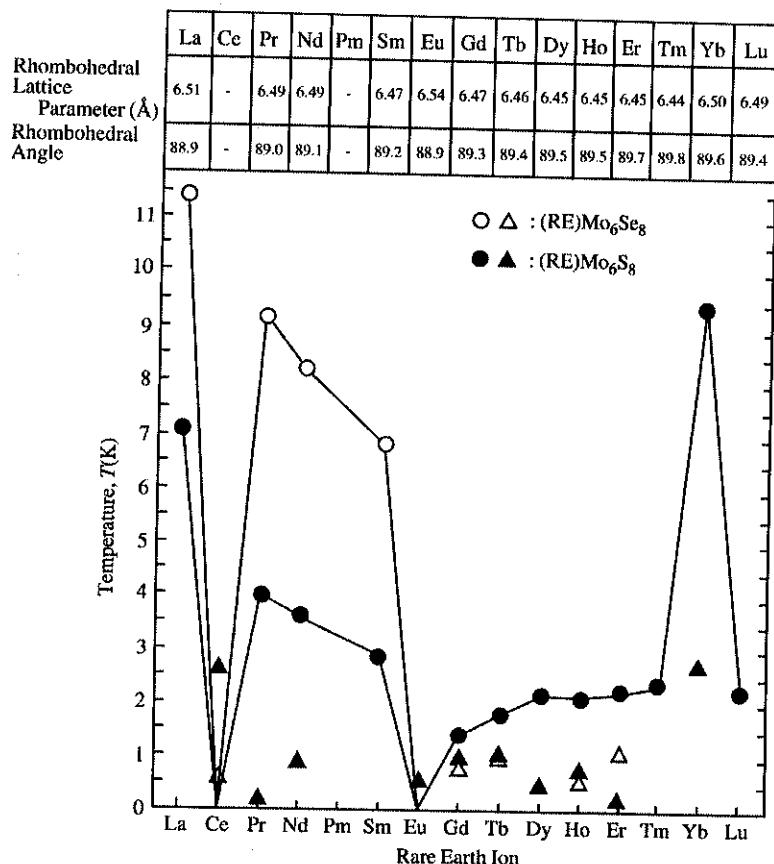
### D1.5.2 Structural properties

The majority of superconducting Chevrel phase superconductors are rhombohedral–hexagonal in structure with space group  $R\bar{3}$ . A schematic presentation of the structure, as well as a simplified cubic structure, is illustrated for  $M_xMo_6X_8$ , where M is Pb and X is S, in figure D1.5.3 [21]. The sides of all the axes of the rhombohedron are equal and inclined at the same angle ( $\sim 90^\circ$ ) to each other and hence can also be considered a slightly distorted cubic structure. The Pb is almost at the centre of the central cube. At the corners of the central cube are the tightly bonded  $Mo_6S_8$  clusters. Mo forms an octahedron such

**Table D.1.5.1.** The Critical temperature of Chevrel phase materials and related compounds: the critical temperature ( $T_c$ ) [1, 2]; the rhombohedral angle, the rhombohedral lattice parameter and the Mo–Mo intracluster spacing [3–5]

Compound	$T_c$	Rhombohedral lattice parameter (Å)	Rhombohedral angle	Mo–Mo intracluster distance
PbMo <sub>6</sub> S <sub>8</sub>	15 K	6.55	89.4	3.27
SnMo <sub>6</sub> S <sub>8</sub>	13 K	6.52	89.7	3.23
AgMo <sub>6</sub> S <sub>8</sub>	9 K	6.48	92.0	3.15
ScMo <sub>6</sub> S <sub>8</sub>	3.6	–	–	–
YMo <sub>6</sub> S <sub>8</sub>	3.0	6.45	89.5	–
VMo <sub>6</sub> S <sub>8</sub>	8.2	–	–	–
NbMo <sub>6</sub> S <sub>8</sub>	3.5	–	–	–
LaMo <sub>6</sub> S <sub>8</sub>	7.1	6.51	88.9	–
Mo <sub>6</sub> S <sub>8</sub>	1.6	6.43	91.3	3.08
Cu <sub>1.8</sub> Mo <sub>6</sub> S <sub>8</sub>	11	6.48	94.9	3.24
C <sub>3.2</sub> Mo <sub>6</sub> S <sub>8</sub>	6.4	–	–	3.34
Cu <sub>4</sub> Mo <sub>6</sub> S <sub>8</sub>	< 1 K	6.59	95.6	3.39
Cd <sub>1.1</sub> Mo <sub>6</sub> S <sub>8</sub>	3.5	6.52	92.8	–
Li <sub>4</sub> Mo <sub>6</sub> S <sub>8</sub>	4.4	6.62	94.5	–
Mg <sub>1.14</sub> Mo <sub>6</sub> S <sub>8</sub>	3.5	6.51	93.6	–
Cu <sub>1.2</sub> Mo <sub>6</sub> S <sub>8</sub>	5.6	–	–	–
Zn <sub>1.1</sub> Mo <sub>6</sub> S <sub>8</sub>	3.6	6.49	94.7	–
Cu <sub>2</sub> Mo <sub>6</sub> S <sub>6</sub> O <sub>2</sub>	9	6.54	95.51	–
PbMo <sub>6</sub> S <sub>8</sub> O <sub>2</sub>	11.7	6.52	88.98	–
PbMo <sub>6</sub> Se <sub>8</sub>	6.7	6.81	89.23	–
SnMo <sub>6</sub> Se <sub>8</sub>	6.8	6.75	89.4	–
AgMo <sub>6</sub> Se <sub>8</sub>	6	6.73	91.7	–
Cu <sub>2.8</sub> Mo <sub>6</sub> Se	6	6.79	94.9	–
Mo <sub>6</sub> Se <sub>8</sub>	6.4	6.79	94.9	–
Mo <sub>6</sub> Se <sub>4.8</sub> Te <sub>3.2</sub>	2.7	–	–	–
Mo <sub>6</sub> Se <sub>7</sub> Br	7.1	–	–	–
Mo <sub>6</sub> S <sub>6</sub> Br <sub>2</sub>	13.8	–	–	–
Mo <sub>6</sub> S <sub>6</sub> I <sub>2</sub>	14.0	–	–	–
Mo <sub>6</sub> Se <sub>7</sub> I	7.6	–	–	–
Mo <sub>6</sub> Te <sub>6</sub> I <sub>2</sub>	2.6	–	–	–
Mo <sub>4</sub> Re <sub>2</sub> Te <sub>8</sub>	3.5	–	–	–
Mo <sub>6</sub> S <sub>4.8</sub> Te <sub>3.2</sub>	2.7	–	–	–

that each Mo atom is slightly outside the middle of the faces of the S-cube. The Mo<sub>6</sub>S<sub>8</sub> is bound together as a cluster with weak intercluster Mo–Mo bonds. The sides of the unit cell are about 6.5 Å and those of the Mo<sub>6</sub>S<sub>8</sub> cluster about 3.8 Å. The 4d orbitals of the Mo ions are well extended, so they favour the metallic bond. The clusters are rotated through ~25° about the ⟨111⟩ and thus form the channels in which the M atoms are located. The bond lengths of the Mo<sub>6</sub>X<sub>8</sub> cluster in ternary Chevrel phase materials are generally similar to those of the binary parent compounds. All the compounds have metal–metal bonds within the cluster. An important structural feature that affects superconducting properties is the Mo–Mo intracluster distance, which can vary from 3.1 to 3.6 Å (cf table D1.5.1), and which is correlated with the number of valence electrons on the Mo<sub>6</sub>S<sub>8</sub> cluster [22]. The ionic-covalent

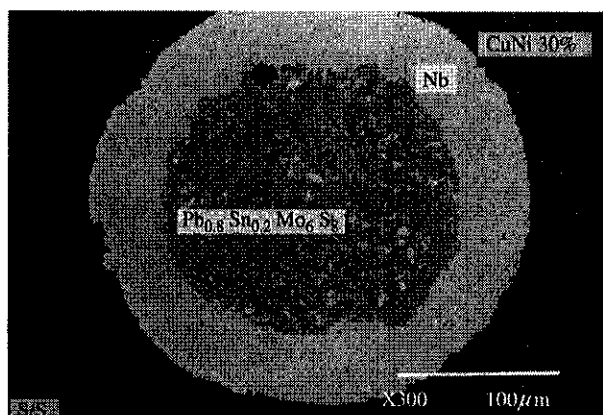


**Figure D1.5.1.** The rare-earth Chevrel phase superconductors: the critical temperature ( $T_c$ :  $\circ$   $\bullet$ ), the magnetic ordering temperatures ( $\Delta$   $\blacktriangle$ ) [1, 2, 6, 7]; the rhombohedral angle and the rhombohedral lattice parameter [5]. No single crystals of heavy rare-earth selenides (i.e. Gd–Yb) have yet been obtained [8] — two phase samples with  $\text{Mo}_6\text{Se}_8$  ( $T_c \sim 6.4$  K) are produced.

cluster–cluster bond can be doped from an insulating state to produce a metallic bond that is superconducting. Chevrel phase compounds can be divided into two types — stoichiometric and non-stoichiometric.

(a) Stoichiometric compounds of  $\text{M}_x\text{Mo}_6\text{X}_8$  contain large cations such as Pb, Sn, Ag and the rare-earth elements. The structure basically accommodates 1 M ion in one of the six positions which are sufficiently close together to appear as a single site. There is a narrow range of M solubility. All stoichiometric compounds crystallize in the rhombohedral structure at high temperatures with the rhombohedral angle between 88 and 90°. A typical x-ray diffraction pattern for  $\text{PbMo}_6\text{S}_8$  is shown in figure D1.5.4 [23]. The positions of the main diffraction lines and  $h$ ,  $k$ ,  $l$  indices for the Chevrel phase compound  $\text{PbMo}_6\text{X}_8$  and the most important second phases are given in table D1.5.2.

(b) Non-stoichiometric compounds contain small ions such as Li, Cu and Zn. The rhombohedral angle is between 92 and 95°. The structure can accommodate more than one ion. Figure D1.5.5 shows the possible sites for Cu in the inner cube of the Chevrel phase structure [5]. There are six inner sites and six outer sites. The sites for the small cations depend on the particular cation. With In for example, only



**Figure D1.5.2.** A  $\text{Pb}_{0.6}\text{Sn}_{0.4}\text{Mo}_6\text{S}_8$  powder-route wire (from [10] and [11]) which has a critical current density value of  $6.7 \times 10^8 \text{ A m}^{-2}$  at 14 T and 4.2 K [12]. Nb is a diffusion barrier. CuNi30% is a hard material that reduces damage to Nb during drawing. Stainless-steel (S/S) provides additional mechanical strength and precompresses the Chevrel phase core after cool-down.

the six inner sites are available. At low temperatures the M ion will freeze in one position which breaks the rhombohedral symmetry and can favour a phase transition to a triclinic structure.

The structural transformation from high temperature rhombohedral to triclinic ( $P\bar{1}$ ) between 100 and 140 K [24], has been observed in  $\text{PbMo}_6\text{S}_8$  [25] using neutron scattering. Synchrotron data show a much smaller triclinic distortion, which suggests that sample preparation (possibly oxygen contamination [26]) affects the low temperature transformation [27]. The triclinic phase is the stable low-temperature phase for the non-superconducting divalent ( $\text{Eu}^{2+}$ ,  $\text{Ba}^{2+}$ ,  $\text{Sr}^{2+}$  and  $\text{Ca}^{2+}$ ) molybdenum sulphide Chevrel phases [28]. Steric and electronic effects (principally charge transfer to the  $\text{Mo}_6\text{S}_8$  cluster) are important in determining the equilibrium structure [8]. Materials with the highest  $T_c$  are those metallic compounds adjacent to a structural instability. They may be in a mixed-phase region consisting of both a superconducting rhombohedral phase and an insulating triclinic phase [28]. The structural instability can result either from changing the cation (chemical pressure) or applying pressure directly [24]. A similar structural instability is found in some of the A15 superconductors which undergo a shear martensitic transformation at some temperature above  $T_c$  [29].

The structure of  $\text{Cu}_x\text{Mo}_6\text{S}_{8-y}$  has been studied in detail below 300 K. Four different low temperature modifications of the rhombohedral phase have been observed [30]. For  $x = 1.2$ ,  $T_c = 5.6 \text{ K}$ , for  $x = 1.8$ ,  $T_c = 11 \text{ K}$ , for  $x = 3.2$ ,  $T_c = 6.4 \text{ K}$  and for  $x = 4$  the material is not superconducting [18]. Less detailed studies have also been completed on Ni molybdenum sulfides and selenides.

Solid solutions of Chevrel phase materials can generally be fabricated if the end compounds exist. The solution can either occur with the chalcogenides (e.g.  $\text{M}_x\text{Mo}_6\text{Se}_{8-x}\text{S}_x$ ) or with the M-elements [17]. Neutron measurements show that in the ternary Pb- and Sn- Chevrel phase sulphides, oxygen can substitute for sulfur which strongly affects the superconducting properties [31]. Of note is the  $\text{EuMo}_6\text{Se}_8$  compound in which either a vacancy or oxygen substitution for Se causes the Eu to move 0.9 Å away from the usual central site [32]. Low level doping of 1–1.5 at% Pb or rare-earth ions into the Chevrel phase binary compound  $\text{Mo}_6\text{Se}_8$  has also been reported [33, 34].

Detailed TEM has been performed on a range of bulk samples of  $(\text{Pb,Gd})\text{Mo}_6\text{S}_8$  and used to show that material can be produced with coherent tilt grain boundaries. Figure D1.5.6 shows typical data demonstrating that in good material the grain boundaries are free of second phase. In some cases, dislocations are observed that occur at regular intervals along the boundary [11]. Detailed HREM on

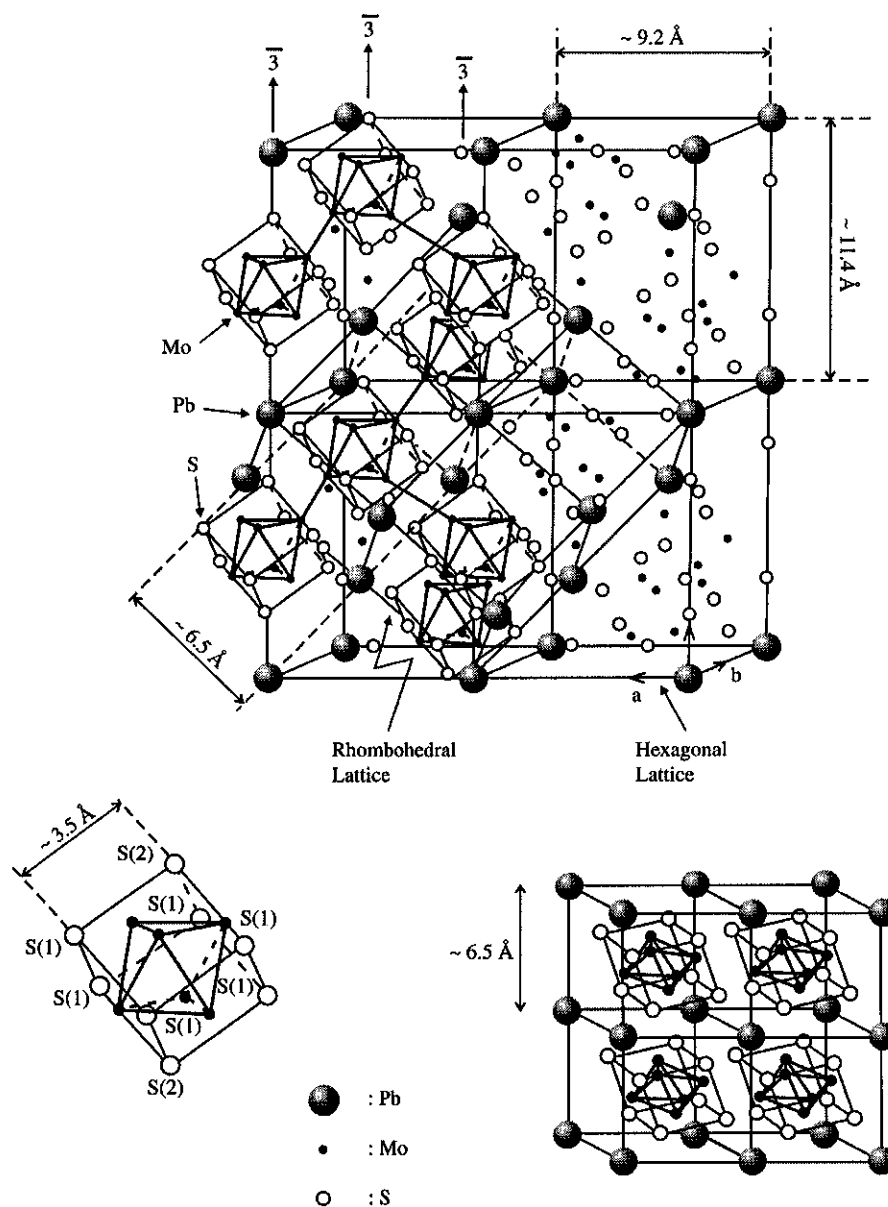
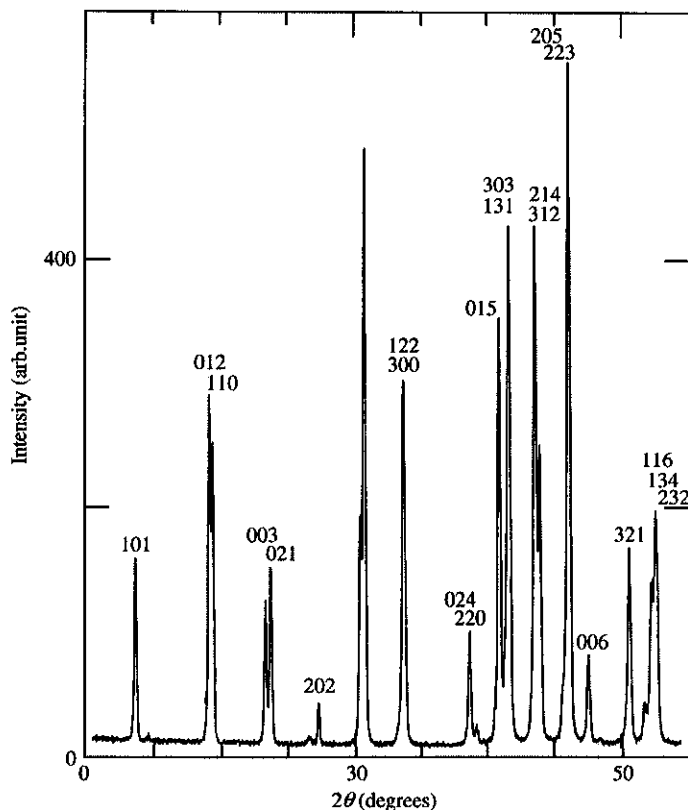


Figure D1.5.3. A schematic of the rhombohedral and cubic structure of  $\text{PbMo}_6\text{S}_8$  [21].

$\text{Ni}_2\text{Mo}_6\text{S}_8$  has shown that edge dislocations, which include an extra plane of  $\text{Mo}_6\text{S}_8$  clusters, can form. Both coherent and incoherent interfaces have been found in this Ni-based compound [35]. A small number of sulfur defects in the clusters were observed (i.e.  $\text{Ni}_2\text{Mo}_6\text{S}_{7.6}$ ).

There is limited work on structures that occur at very high pressures and temperatures (30–80 kbar and  $1200^\circ\text{C}$ ). Preliminary work suggests that a metastable structure can be formed in the  $\text{PbMo}_6\text{S}_8$  system [36] at high pressure which has a  $T_c$  enhanced by about 1 K but a severely reduced upper critical field ( $\sim 7 \text{ T}$ ).



**Figure D1.5.4.** CuK $\alpha$  powder x-ray diffraction powder pattern for a typical single phase PbMo<sub>6</sub>S<sub>8</sub> sample [23].

### D1.5.3 Thermal properties

The specific heat capacity ( $C_P$ ) of PbMo<sub>6</sub>S<sub>8</sub> has been measured by a number of authors in high magnetic fields up to 24 T [37–39]. The electronic contribution in the normal state is  $\sim 7 \text{ mJ K}^{-2}(\text{g atom})^{-1}$ . In the superconducting state, in addition to the BCS exponential gap term, there is also the term linear in temperature that accounts for the normal cores of the fluxons in the mixed state. At the superconducting jump  $C/T$  is  $\sim 70 \text{ mJ K}^{-2}(\text{g atom})^{-1}$  which is relatively high because of the soft modes present [40] and  $\Delta C_e/T_c$  is about  $12 \text{ mJ K}^{-2}(\text{g atom})^{-1}$ . The effective Debye temperature measured using  $C_P$  measurements changes by a factor of 2 from  $\sim 200 \text{ K}$  at 4.2 K up to  $\sim 400 \text{ K}$  at room temperature [41]. Figure D1.5.7 shows a Debye plot for a series of compounds of the form  $\text{Pb}_{1-x}\text{Cu}_{1.8x}\text{Mo}_6\text{S}_8$  [23]. Ultrasonic measurements give an average Debye temperature of  $\sim 245 \text{ K}$  [26]. Many magnetic Chevrel phase superconductors have been measured [4]. Such work includes investigating pressure-induced re-entrant superconductivity [42] and the reduction in  $\Delta C_e/T_c$  with increased magnetic doping [43]. Nevertheless, there is typically a factor of 2 variation in all the parameters derived from specific heat data in the literature (N.B. for PbMo<sub>6</sub>S<sub>8</sub>, 1 mole = 1037 g = 15 g atom). The differences are attributed to the sensitivity of the materials to the fabrication process.

The consensus on the phonon density of states in Chevrel phase materials is good. The weighted phonon density of states for PbMo<sub>6</sub>S<sub>8</sub> and SnMo<sub>6</sub>S<sub>8</sub> measured using neutron scattering measurements and specific heat measurements is consistent with calculations that assume the Mo<sub>6</sub>S<sub>8</sub> clusters are tightly bound but only weakly interact with other clusters or the M ion. There is a relatively flat dispersion curve

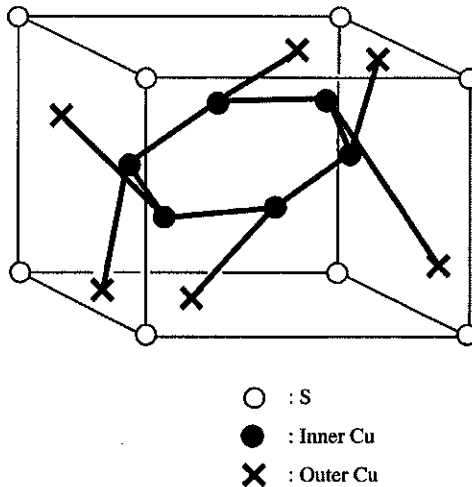
**Table D1.5.2.** The x-ray diffraction lines for  $\text{PbMo}_6\text{S}_8$  and the strong intensity lines ( $> 10\%$ ) for the important impurity phases (Mo,  $\text{MoS}_2$ ,  $\text{Mo}_2\text{S}_3$ , Pb, PbS and S). Radiation:  $\text{CuK}\alpha 1$ ,  $\lambda = 1.54056 \text{ \AA}$  (taken from the International centre for Diffraction data)

2-theta	Int.	<i>h k l</i>	2-theta	Int.	<i>h k l</i>	2-theta	Int.	<i>h k l</i>
<b>PbMo<sub>6</sub>S<sub>8</sub></b>			<b>MoS<sub>2</sub></b>			<b>Pb</b>		
13.612	50	1 0 1	14.38	100	0 0 2	29.55	100	1 0 0
19.153	100	0 1 2	32.67	22	1 0 0	33.66	80	1 0 1
19.364	80	1 1 0	33.51	12	1 0 1	42.82	75	–
23.303	50	0 0 3	35.87	10	1 0 2	52.23	75	1 1 0
23.700	40	0 2 1	39.53	58	1 0 3	57.95	75	1 0 3
–	–	2 0 2	44.15	11	0 0 6			<b>PbS</b>
–	–	1 1 3	49.78	29	1 0 5	25.96	84	1 1 1
30.753	100	2 1 1		<b>Mo<sub>2</sub>S<sub>3</sub></b>		30.07	100	2 0 0
–	–	1 2 2	10.48	10	0 0 1	43.06	57	2 2 0
33.666	80	3 0 0	16.28	95	1 0 1	50.97	35	3 1 1
38.730	20	0 2 4	21.06	36	0 0 2	53.41	16	2 2 2
–	–	2 2 0	29.61	35	$\bar{2}$ 0 1			<b>S</b>
40.971	40	0 1 5	29.73	26	0 1 1	16.59	10	1 2 1
–	–	3 0 3	31.66	46	1 1 0	19.62	17	2 1 2
–	–	1 3 1	31.66	46	0 0 3	22.08	17	2 2 0
–	–	2 1 4	32.33	16	$\bar{1}$ 1 1	23.28	100	2 2 2
43.938	40	3 1 2	32.89	12	$\bar{2}$ 0 2	23.58	23	1 3 2
–	–	2 0 5	34.46	19	1 1 1	24.30	25	1 2 5
–	–	2 2 3	35.09	37	0 1 2	25.37	23	1 3 3
–	–	0 0 6	36.39	11	$\bar{1}$ 1 2	26.05	49	0 2 6
–	–	3 2 1	39.04	28	$\bar{2}$ 0 3	26.84	34	3 1 1
–	–	1 1 6	40.16	58	1 1 2	27.87	36	2 0 6
–	–	1 3 4	40.72	39	2 0 2	28.86	36	1 3 5
–	–	2 3 2	41.03	24	$\bar{2}$ 1 1	31.60	24	0 4 4
	<b>Mo</b>		42.27	21	$\bar{2}$ 0 4	34.27	13	4 0 0
40.51	100	1 1 0	42.93	100	0 1 3	34.37	13	1 3 7
58.60	16	2 0 0	42.93	100	0 0 4	35.06	11	3 3 3
			43.58	51	$\bar{2}$ 1 2	37.17	11	4 0 4
			46.18	23	$\bar{3}$ 0 2	42.84	12	3 1 9

and a strong peak at about 5 meV, associated with the Einstein mode from the M ion. Modes in the energy range up to 18 meV are associated with the soft external modes of the  $\text{Mo}_6\text{S}_8$  clusters. The hard internal modes are responsible for the energy range from 18 up to 50 meV [40].

#### D1.5.4 Mechanical properties

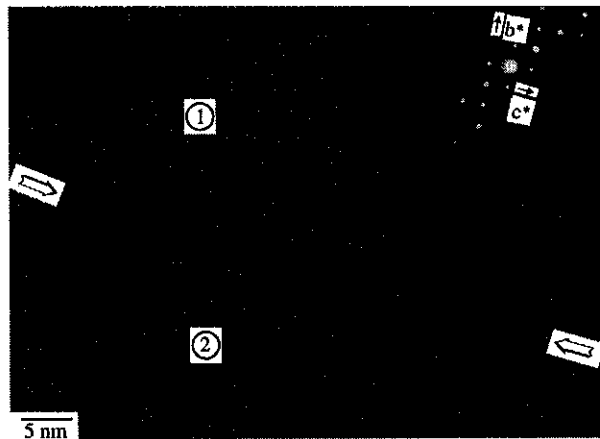
Among Chevrel phase materials, the coefficient of thermal expansion ( $\alpha$ ) has been studied most extensively for  $\text{PbMo}_6\text{S}_8$ . Using x-ray diffraction in the range 10–1200 K on bulk material, it has been concluded that  $\alpha$  is almost temperature independent up to 900 K (650°C), where  $\alpha = 1/L (dL/dT) = 9.4 \times 10^{-6} \text{ K}^{-1}$  [44]. For comparison, at room temperature  $\alpha$  for  $\text{PbMo}_6\text{S}_8$  is about half that of Cu or steel, about 25% higher than Nb and about twice that of Mo [44]. Single crystal measurements show



**Figure D1.5.5.** The possible sites for Cu in the inner cube of  $\text{Cu}_x\text{Mo}_6\text{S}_8$  [5].

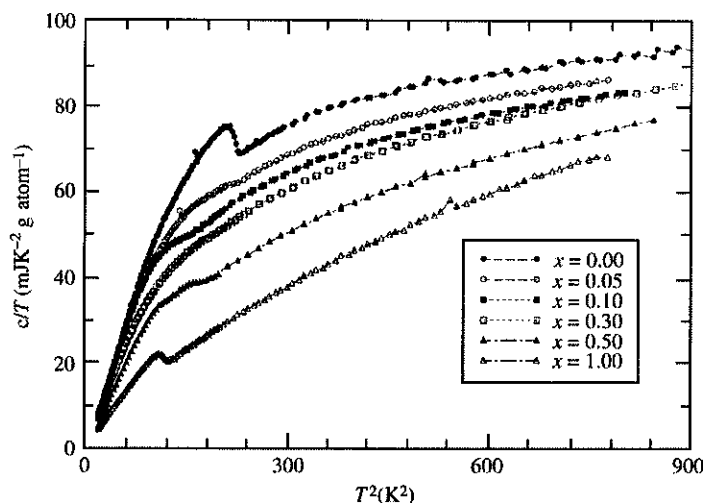
a relatively strong anisotropic variation for  $\alpha$  of about a factor 3 [45]. Such considerations are particularly important for optimizing Chevrel phase wires [46]. High resolution thermal expansion measurements have been made using capacitive techniques. These show a very strong change in  $\alpha$  at  $T_c$  in  $\text{PbMo}_6\text{S}_8$  but not in  $\text{SnMo}_6\text{S}_8$  [47] which was attributed to different coupling between the superconductivity and the orthorhombic–triclinic structural transition (cf chapter B3.3.5).

Compressibilities have been measured for 11 sulfur and selenide Chevrel phase compounds [48] and typical data shown in figure D1.5.8. The Young's modulus calculated from these data is around 40 GPa [46]. This makes it similar to indium, about a factor of 5 smaller than steel and eight times smaller than Mo. Elastic constants have also been measured using ultrasonic techniques. The elastic constant for transverse distortions is 90 GPa and for longitudinal distortions is 21 GPa [26].



**Figure D1.5.6.** A high resolution electron micrograph (HREM) of a grain boundary of  $\text{Pb}_{0.7}\text{Gd}_{0.3}\text{Mo}_6\text{S}_8$  bulk sample. The selected area diffraction pattern (SADP) is for grain (1) and is close to the [100] zone axis. The grain boundary is very narrow [11].



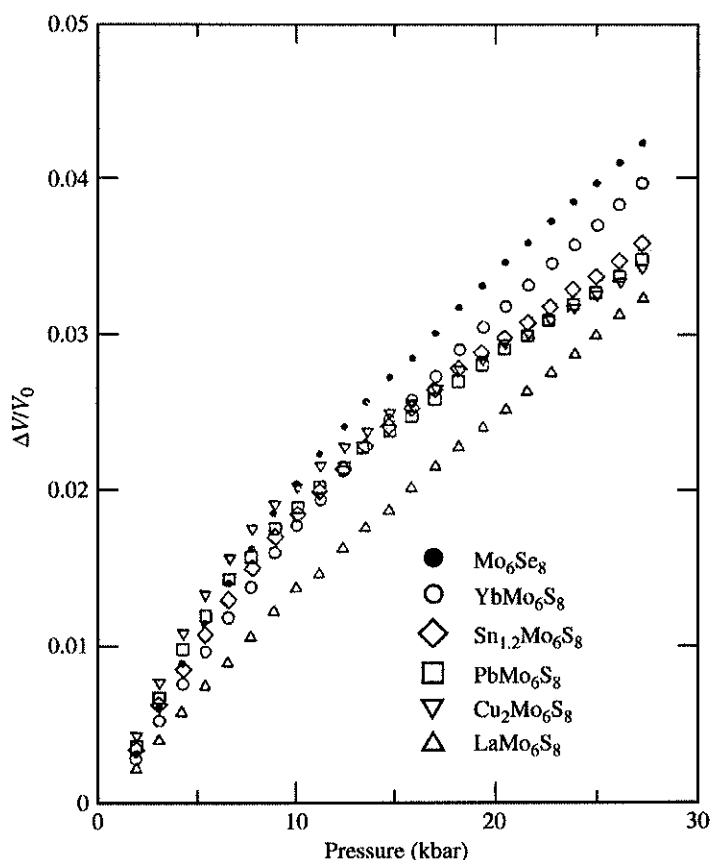


**Figure D1.5.7.** Debye plot [(heat capacity)<sup>2</sup>/(temperature)] versus temperature] for  $(\text{Pb}_{1-x}\text{Cu}_x)\text{Mo}_6\text{S}_8$  for different values of  $x$  [23].

The strain tolerance of Chevrel phase material has been investigated most comprehensively in the context of wires [49].  $\text{PbMo}_6\text{S}_8$  and  $\text{SnMo}_6\text{S}_8$  fracture at about 0.65%, which is typical for a ceramic material. Mechanical properties such as fracture toughness, crack propagation and fatigue properties are strongly dependent on the porosity of the material. Chevrel phase materials, therefore, can be considered as soft ceramics.

### D1.5.5 Chemical properties/phase diagrams

The high vapour pressure of the Pb and S makes accurate phase studies difficult. Very careful exclusion of oxygen and water during material fabrication is required to avoid oxygen substituting for sulfur in the  $\text{Mo}_6\text{S}_8$  cluster [50] and is necessary for reliable comparisons between structure/composition and superconducting properties. Argon should be used rather than argon–nitrogen since trapped nitrogen can form MoN. Chevrel phase materials start decomposing above 650°C [44]. Most of the important compounds melt in the temperature range from 1500 to 2000°C and the vapour pressure of S or Se (Pb, Sn) is high [7, 51]. Pb, Sn, Ag and rare-earth sulfide and selenide Chevrel phase materials melt peritectically [3]. To make single crystals of the metallic superconductors, a typical off-stoichiometric composition of  $\text{Pb}_{1.2}\text{Mo}_7\text{S}_8$  [30] is used and for the rare-earth compounds,  $\text{RE}_{11}\text{Mo}_{38}\text{S}_{51}$  [7, 52] and  $\text{RE}_{15}\text{Mo}_{30}\text{Se}_{55}$  [8]. These compositions are chosen to produce excess binary chalcogenide which minimizes formation of the  $\text{Mo}_2\text{S}_3$  phase which competes with the Chevrel phase. Figure D1.5.9 (upper) shows a schematic of the phase diagram for  $\text{PbMo}_6\text{S}_8$  found at  $\sim 1000^\circ\text{C}$  [53, 54]. The  $\text{Mo}_2\text{S}_3$  phase does not form at lower temperatures [55]. The high temperature phase diagram characteristic of the rare-earth compounds is shown in figure D1.5.9 (lower) [7]. The material can be simply cooled from above its melting point to produce single crystals. The final product is generally single crystals in a binary chalcogenide crust which can be separated using HCl diluted with ethyl alcohol ( $\approx 20$  vol% HCl). The natural cleavage planes are the (100) and (110) crystallographic directions [56]. The Cu and Ni sulphide Chevrel phase materials form congruently so that relatively large single crystals can be formed from a stoichiometric melt. In selenium based materials, only single crystals of the light rare-earths have been produced. In heavy

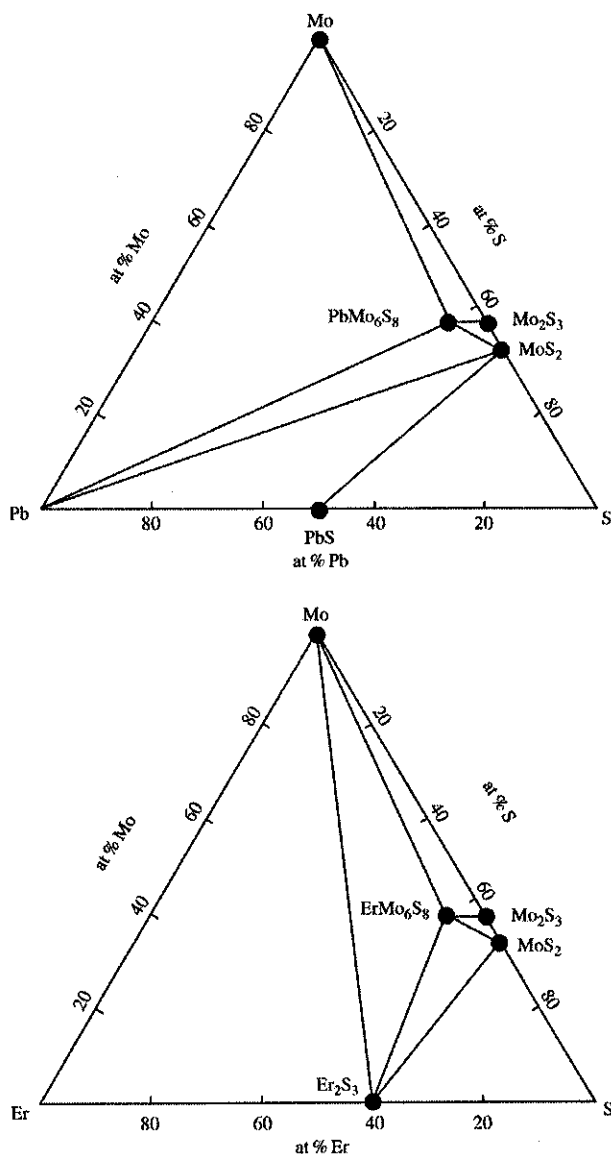


**Figure D1.5.8.** The compressibility of five ternary molybdenum sulphides  $M_x\text{Mo}_6\text{S}_8$  and the binary compound  $\text{Mo}_6\text{Se}_8$  [48].

rare-earth selenides, RE deficiency produces a two phase sandwich structure of doped binary  $\text{Mo}_6\text{Se}_8$  and the  $\text{REMo}_6\text{Se}_8$  [8, 57–59].

A number of authors have studied the phase diagrams of the  $\text{Pb}_x\text{Mo}_6\text{S}_{8-y}$  at 900 [55] and  $\sim 1000^\circ\text{C}$  [53, 54]. There is no general agreement about the composition of the Chevrel phase material in the single phase region although many studies show the presence of a small amount of sulfur defects, consistent with HREM studies [35]. Recently it has been concluded [60] that as long as there is no oxygen contamination, the onset of  $T_c$  is in the range of 14–15 K, the sulfur stoichiometry is very close to 8 and Pb deficiency is present. Chevrel phase materials have been fabricated at high pressure (typically up to 2000 atm) to increase density and connectivity. It has been suggested that this enhances the effect of oxygen contamination since the material is pushed into a two-phase Chevrel phase +  $\text{MoS}_2$  region [47]. Limited work has been completed on decomposition. Since  $\text{PbMo}_6\text{S}_8$  can be formed between 450 and  $1650^\circ\text{C}$ , it is reasonable to assume that the structure is very stable. Electron-beam-induced decomposition has been observed in  $\text{Ni}_2\text{Mo}_6\text{S}_8$  first showing increased disorder and then, after time, reduction of the  $\text{Mo}_6\text{S}_8$  clusters giving regions of Mo, Mo–Ni alloy and sulfides [35].

Fine grain, well connected bulk samples are required for high  $J_c$  applications. Soft chemistry methods can be used to produce ultrafine precursors ( $\text{PbS}$ ,  $\text{MoS}_2$  and Mo) [61]. Alternatively very fine grains of the binary compound  $\text{Mo}_6\text{S}_8$ , which are almost oxygen free, can be formed by leaching the Ni



**Figure D1.5.9.** A schematic of the phase diagram for  $\text{PbMo}_6\text{S}_8$  (upper) [53, 54] and  $\text{ErMo}_6\text{S}_8$  (lower) at temperatures above  $1000^\circ\text{C}$  [7].

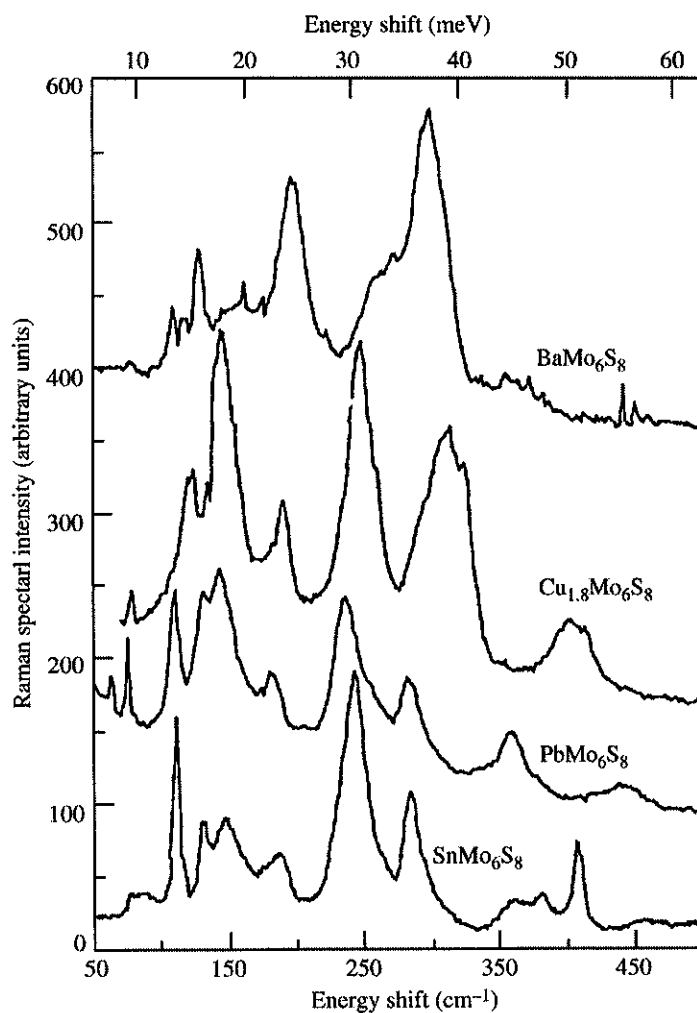
or Li ions out of the parent Chevrel phase compound using HCl [62, 63]. In this remarkable process, the Ni or Li ions can migrate distances of  $\sim 100 \mu\text{m}$  (with potential uses for battery applications). The fine  $\text{Mo}_6\text{S}_8$  grains can then be used in powder route fabrication of fine grain bulk Chevrel phase materials by reacting with PbS or SnS [64, 65].

Substitutions of Br, I and O for the X element and mixed sulfur–selenium compounds have been fabricated. The Br, I, and O substitutions can increase  $T_c$ . Replacing S by Se or vice-versa immediately reduces  $T_c$  producing a minimum when there are equal quantities of each element (e.g.  $\text{PbMo}_6\text{Se}_4\text{S}_4$ ).

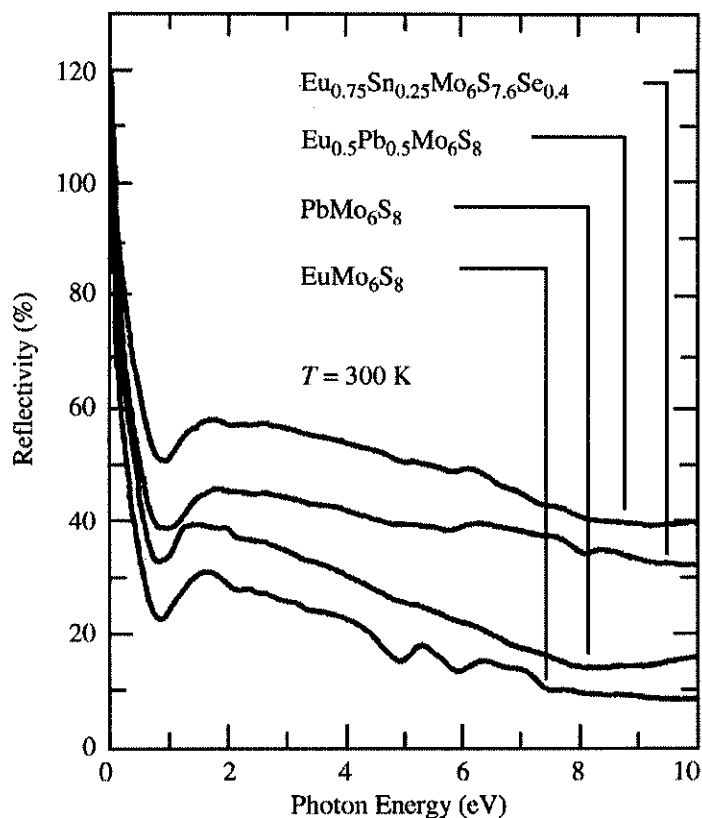
Most materials in which the Mo element has been substituted (e.g. Nb, Ta, Re, Ru and Rh) are not superconducting [5]. A detailed investigation of the phase diagram at 1200°C in La–Mo–Se has also been completed and the correlation between structural and superconducting properties investigated [57, 66, 67].

### D1.5.6 Optical properties

Vibrational Raman spectra have been observed for Cu-, Pb-, Ba- and Sn sulfide Chevrel phase compounds in the range from 10–50 meV [56] as shown in figure D1.5.10. Such measurements give energies and symmetries of the Raman active optical phonons near the Brillouin zero centre. Many peaks are independent of the metal (X) atom even in the non-stoichiometric Cu- compounds, in broad agreement with tunnelling data [68].



**Figure D1.5.10.** The Raman spectra for BaMo<sub>6</sub>S<sub>8</sub>, Cu<sub>1.8</sub>Mo<sub>6</sub>S<sub>8</sub>, PbMo<sub>6</sub>S<sub>8</sub> and SnMo<sub>6</sub>S<sub>8</sub> [56].



**Figure D1.5.11.** The near normal reflectivity at room temperature of  $\text{EuMo}_6\text{S}_8$ ,  $\text{PbMo}_6\text{S}_8$ ,  $\text{Pb}_{0.5}\text{Eu}_{0.5}\text{Mo}_6\text{S}_8$  and  $\text{Sn}_{0.25}\text{Eu}_{0.75}\text{Mo}_6\text{S}_{7.6}\text{Se}_{0.4}$  [69].

Reflectivity measurements and complex magneto-optical Kerr-effect measurements have been completed in magnetic fields up to 12 T and temperatures down to 0.5 K [69]. Figure D1.5.11 shows that the reflectivity (and optical conductivity spectra) were similar for  $\text{Eu}_{1-x}\text{Pb}_x\text{Mo}_6\text{S}_8$  and  $\text{Eu}_{1-x}\text{Sn}_x\text{Mo}_6\text{S}_{8-y}\text{Se}_y$  at 300 K. The carrier density was low ( $\sim 6 \times 10^{27} \text{ m}^{-3}$ ), the mobility  $\sim 1 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$  and the effective masses for the carriers  $\sim 10 m_e$ .

### D1.5.7 Normal state properties

The normal state properties of Chevrel phase compounds are rather well described in terms of their chemistry and are consistent with band structure calculations [70]. The dominant carriers are holes associated with the  $4d_{x^2-y^2}$  states. The three important factors that contribute to the electronic properties are the charge transfer from the M cation to the  $\text{Mo}_6\text{S}_8$  cluster and the volume and structure of the unit cell.

The position of the Fermi level in the sulfide system, is strongly related to the charge transfer between the M cations and the sulfur anions [22]. A characteristic of the  $\text{Mo}_6\text{S}_8$  cluster is that it is only slightly distorted when filled with 24 electrons and tends to be insulating. The electronic configuration of  $\text{Mo}4d^5 5s^1$  contributes six electrons to the  $\text{Mo}_6\text{S}_8$  cluster. Hence with the S in the  $-2$  valence state, there

are 20 electrons in the  $\text{Mo}_6\text{S}_8$  cluster and there is hole conduction. Adding Pb or Sn to the structure (both have valence +2) increases the number of electrons to 22—equivalent to two holes per cluster which produces the highest value of  $T_c$ . For comparison the valence of S, Se and Te in  $\text{Mo}_6\text{S}_8$ ,  $\text{Mo}_6\text{Se}_8$  and  $\text{Mo}_6\text{Te}_8$  are  $-2$ ,  $-1.75$  and  $-1.33$ , respectively. Band structure calculations show that in the rhombohedral  $\text{PbMo}_6\text{S}_8$ , the Fermi energy lies below an energy gap about 1 eV wide [70]. The states near the Fermi energy are strongly confined within the  $\text{Mo}_6\text{S}_8$  cluster [71]. Indeed there are some broad similarities between Chevrel phase materials and the intercalates of  $\text{TaS}_2$  [72] and  $\text{MoS}_2$  [73] where there is a hybridization of the metal d-bands and an associated energy gap [74]. The band structures for the Chevrel phase selenides and the telluride have similar properties [71], but one of the marked differences between them and the sulfides is that the density of states at the Fermi energy is higher for trivalent ions than for divalent ions. This is supported by the  $T_c$  values which are higher for the rare-earth selenides than for the sulphides as shown in figure D1.5.1.

In  $\text{PbMo}_6\text{S}_8$  samples, the resistivity ( $\rho$ ) at room temperature is about  $100 \mu\Omega \text{ cm} - 1 \text{ m}\Omega \text{ cm}$  and typical values for room-temperature-resistivity-ratio (RRR) are 4–6 [23, 44]. In the  $\text{Cu}_x\text{Mo}_6\text{S}_8$  single crystals,  $\rho$  is similar with a RRR value of about 7 which leads to a scattering length ( $l$ ) of about 20–30 Å [3]. In thin films, RRR values were found in the range 2–6 and  $l$  estimated to be  $\sim 40$  Å [75]. The temperature dependence of  $\rho(T)$  is approximately linear up to 50 K but shows negative curvature at higher temperatures similar to the A15 superconductors. Theories that address the non-linearity utilize a strong peak in the density of states [76] or a scattering length that is comparable to the lattice spacing [77]. Experimental data at low temperatures can be misleading, particularly if the material includes pure Mo or  $\text{Mo}_2\text{S}_3$ . Hall effect measurements in the  $(\text{Eu}_{1-x})\text{Sn}_x\text{Mo}_6\text{S}_8$  at room temperature gives Hall coefficients of  $+0.7 \times 10^{-3} \text{ cm}^3 \text{ G}^{-1}$  confirming a hole carrier concentration of  $\sim 9 \times 10^{27} \text{ m}^{-3}$  [78] (in agreement with the optical measurements and muon measurements [79]) and implying  $\sim 2.5$  holes per formula unit. In  $\text{REMo}_6\text{S}_8$  compounds,  $\rho$  is typically  $\sim 300 \mu\Omega \text{ cm}$  and RRRs range from 8 to 34 [80], although  $\text{LuMo}_6\text{S}_8$  has a resistivity at room temperature of only  $50 \mu\Omega \text{ cm}$  [81]. Limited thermopower measurements on  $\text{Cu}_{1.8}\text{Mo}_6\text{S}_{8-x}\text{Te}_x$  have also been completed [82].

The density of states derived from susceptibility measurements, specific heat measurements and band structure calculations give consistent values. For the Pb- and Sn- Chevrel phase sulfide compounds,  $\chi$  is about  $3.5 \times 10^{-5} \text{ emu}(\text{g atom})^{-1}$ . The (phonon-enhanced) density of states calculated from  $C_P$  measurements is about 1 state per eV-atom-spin. This is about twice that found from  $\chi$  measurements or bandstructure calculations which is expected with strong electron–phonon coupling [17]. In materials with relatively high critical temperature  $\text{PbMo}_6\text{S}_8$ ,  $\text{LaMo}_6\text{S}_8$  and  $\text{LaMo}_6\text{Se}_8$ , the susceptibility varies by about a factor of 1.5–2 between room temperature and 20 K [7, 67] and shows an anisotropy (for single crystals) of  $\sim 40\%$ . A strongly temperature dependent susceptibility has also been observed in high temperature superconductors. This is taken to be evidence that the Fermi level is situated near a peak in the density of states which may cause the relatively high  $T_c$  values.

## D1.5.8 Superconducting properties

### D1.5.8.1 Transition temperature

The theory by Bardeen, Cooper and Schrieffer (BCS theory) [19, 83] currently provides the only generally accepted microscopic explanation for superconductivity. Within this framework, the critical temperature is determined by the density of states at the Fermi level, the phonon spectrum and the electron–phonon coupling. Tunnelling measurements on  $\text{Cu}_{1.8}\text{Mo}_6\text{S}_8$  and  $\text{PbMo}_6\text{S}_8$  give the ratio of the gap ( $\Delta$ ) to  $T_c$  of  $2\Delta/k_B T_c = 4 - 5$  (BCS theory predicts 3.5) showing strong coupling. In the pseudobinary  $\text{Mo}_6\text{Se}_{1-x}\text{S}_x$  system, strong coupling in  $\text{Mo}_6\text{Se}_8$  ( $T_c = 6.2 \text{ K}$ , coupling constant:  $\lambda = 1.25$ ,  $2\Delta/k_B T_c = 4.2$ ,  $\Delta C_c/C_c = 2.25$ ) gives way to weak coupling showing the BCS behaviour in  $\text{Mo}_6\text{Se}_4\text{S}_4$  ( $T_c = 1.8 \text{ K}$ ,

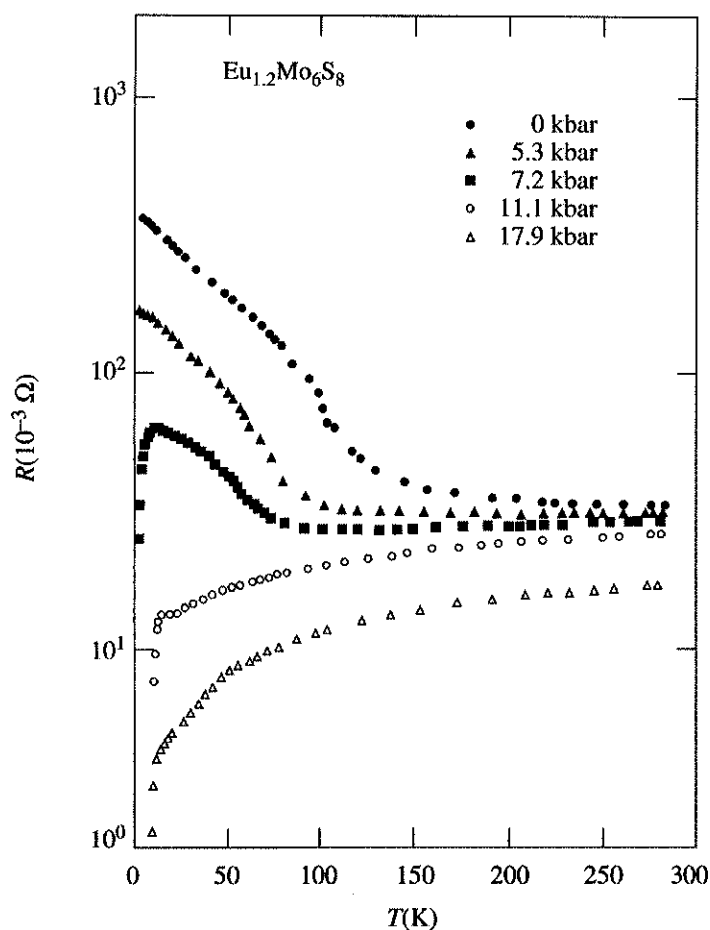
$\lambda = 0.6$ ,  $\Delta C_e/C_e = 1.4$ ) [84, 85]. Although most of the phonon modes associated with the  $\text{Mo}_6\text{S}_8$  cluster modes contribute to the electron coupling [86], the higher  $T_c$  materials have large values of  $\lambda$  which are probably most strongly affected by the soft modes [84]. The pseudobinary  $\text{Mo}_6\text{S}_6\text{I}_2$  compound has the relatively high  $T_c$  of 14 K which suggests that the superconductivity in Chevrel phases of highest  $T_c$  is fundamentally associated with the clusters. In this context, the isotope effect observed in  $\text{Mo}_6\text{Se}_8$  is consistent with BCS theory and suggests that an electron–phonon mechanism operates in Chevrel phase materials [85]. For a given structure,  $T_c$  also depends on the volume of the unit cell [50] and the valence electron concentration in the  $\text{Mo}_6\text{S}_8$  cluster. The trivalent sulfur-based Chevrel phase materials have uniformly low  $T_c$  whereas the divalent rhombohedral materials have high  $T_c$  (e.g. Sn and Pb). Among the trivalent rare-earth ions, there is a correlation between  $T_c$  and the volume of the unit cell [17]. When the volume decreases, the inter-cluster Mo–Mo decreases, so the valence bands are expected to broaden, and the density of states and  $T_c$  to fall [2]. Hydrostatic pressure has been used to change the volume of the unit cell, and hence  $T_c$ , in divalent sulphides [50, 87, 88]. The effect of pressure on  $T_c$  is about an order of magnitude higher than that found in elemental superconductors [89]  $dT_c/dP \sim 10^{-4} \text{ K bar}^{-1}$ . The difference in critical temperature ( $T_c$ ) between  $\text{PbMo}_6\text{S}_8$  and  $\text{SnMo}_6\text{S}_8$  can be explained by the difference in the volume of the unit cell.

The Chevrel phase materials that transform fully from a rhombohedral structure at high temperatures to a triclinic structure at low temperatures have low electronic density of states [90] and are non-superconducting. However, the superconductivity can be restored if pressure is applied to prevent the triclinic structure occurring. In  $\text{BaMo}_6\text{S}_8$  for example, applying a pressure of 4 GPa changes the material from a triclinic semiconductor to a mixed triclinic–rhombohedral phase that is metallic with a  $T_c$  of 12 K [91]. The structural instabilities in the Chevrel phase superconductors may enhance the electron–phonon coupling. In  $\text{Eu}_{1.2}\text{Mo}_6\text{S}_8$  [92], (shown in figure D1.5.12) superconductivity close to the metal–insulator transition is tuned by pressure. In  $(\text{Sn}_{1-x}\text{Eu}_x\text{M})_{1.2}\text{Mo}_6\text{S}_8$  [93] the transition is tuned by Sn content (or carrier concentration). These properties are reminiscent of the HTS materials where high  $T_c$  values also occur in materials with relatively low carrier concentration which are in proximity to the metal–insulator transition. There has been intense research into alternative microscopic mechanisms for superconductivity following the discovery of the high temperature superconductors in the late 1980s. There is some empirical evidence that the cuprate and bismuthate high temperature superconductors, the organic, Chevrel-phase and heavy Fermion systems all belong to a single class of superconductors where  $T_c$  is proportional to the (small)  $n_s/m^*$  (carrier density/effective mass) [20, 94] as shown in figure D1.5.13. In the cuprates, the high values of  $T_c$  and the lack of a clear isotope effect suggest a non-phononic mechanism [95]. This suggests that Chevrel phase materials, which have a well-known chemistry and electronic structure, may be model systems in which to investigate non-standard mechanisms for superconductivity because of the simplifications which follow from their (almost) cubic (isotropic) structure. Alternatively, there is evidence that this simple empirical proportionality shown in figure D1.5.13 does not hold in Chevrel phase materials [79] but is best described using a rather more complex percolation model dependence [96]. The microscopic mechanism for superconductivity in the Chevrel phase materials had long been considered classic BCS because of the isotope effect found in  $\text{Mo}_6\text{Se}_8$ , although this is now no longer a closed issue.

### D1.5.8.2 Upper critical field

The upper critical field ( $B_{c2}$ ) is given by

$$B_{c2}(0) \approx \left[ 8.3 \times 10^{34} \left( \frac{\gamma T_c}{S} \right)^2 + 3.1 \times 10^3 \gamma \rho_N T_c \right] \quad (\text{D1.5.1})$$



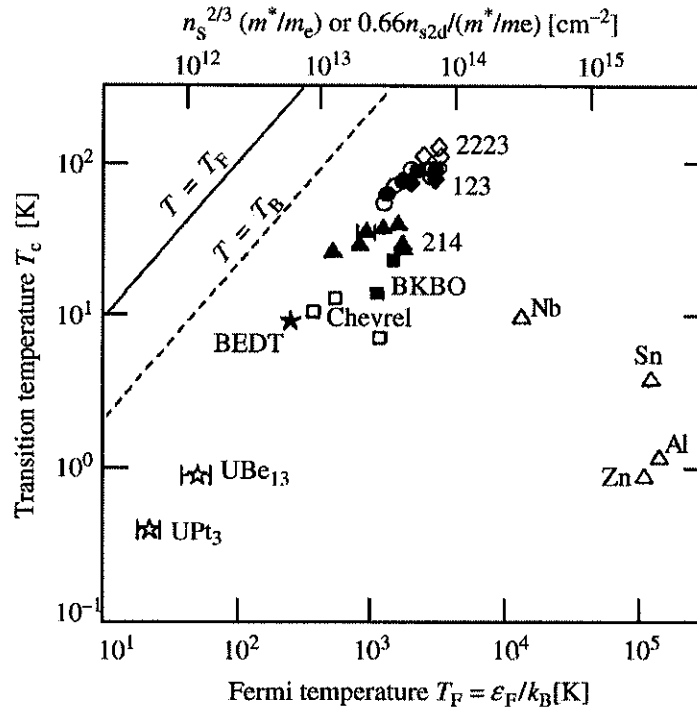
**Figure D1.5.12.** The resistance versus temperature for  $\text{Eu}_{1.2}\text{Mo}_6\text{S}_8$  at various pressures [92].

where the two terms are the clean and dirty contributions, respectively [60, 97], and are consistent with  $\xi_0 \sim l$ . Both terms contribute to the very high upper critical field values in the  $\text{PbMo}_6\text{S}_8$  and  $\text{SnMo}_6\text{S}_8$  compounds. The effect of the intrinsic spin, orbital coupling and spin-orbit coupling must be included in the Werthamer–Helfand–Hohenberg (WHH) theory to describe the temperature dependence of  $B_{c2}$  [98]. However, further theoretical work is still required to assess whether the parameters derived are physically significant. The anisotropy of  $B_{c2}$  in  $\text{PbMo}_6\text{S}_8$ ,  $\text{PbMo}_6\text{Se}_8$ ,  $\text{Cu}_{1.8}\text{Mo}_6\text{S}_8$  and  $\text{SnMo}_6\text{Se}_8$  has been found experimentally to be about 15% [99–101] (as shown in figure D1.5.14) and correlated with the rhombohedral angle (cf Chapter B3.3.5). There is currently no adequate explanation for this since the band structure calculations show nearly cubic symmetry and predict low anisotropy.

### D1.5.8.3 Ginzburg–Landau description

Ginzburg–Landau (G–L) theory provides a self-consistent explanation for the properties of metallic superconductors in-field (i.e. superconductors that are non-magnetic in the normal state). There are only two free parameters which can be taken to be the G–L constant ( $\kappa$ ) which is temperature independent





**Figure D1.5.13.** A log–log plot of critical temperature ( $T_c$ ) versus the Fermi temperature ( $T_F$ ) estimated from muon spin resonance measurements (combined with the interplanar distance for two-dimensional and the Sommerfeld constant for three-dimensional systems) for cuprates, BKBO, Chevrel phase, BEDT, heavy-fermion and some elemental superconductors. The dashed line represents the Bose–Einstein condensation temperature,  $T_B$ , of the ideal boson gas [20].

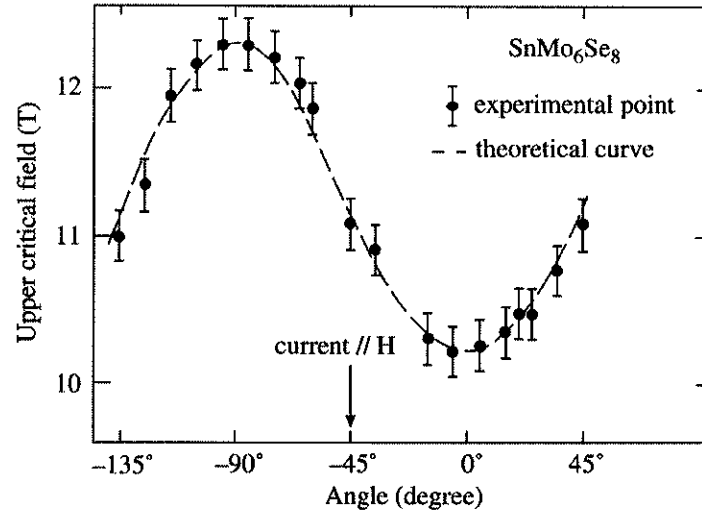
and  $B_{c2}(T)$  [102]. The fundamental properties of the superconducting state can be determined by measuring the reversible magnetization close to  $B_{c2}$  and using the G–L relation:

$$M = \frac{-(H_{c2} - H)}{(2k^2 - 1)\beta_A} \quad (\text{D1.5.2})$$

where  $H_{c2}$  is the critical field strength,  $H$  is the applied field strength and  $\beta_A$  is the Abrikosov constant. Reversible magnetization measurements similar to those shown in figure D1.5.15, can be used to obtain values for  $k$  and  $dB_{c2}/dT$ , from which the slopes  $dB_c/dT$  and  $dB_{c1}/dT$  can be calculated. For high  $\kappa$  materials such as the Chevrel phase superconductors, it is best not to calculate the critical fields at low temperatures using the G–L relations directly, since G–L theory is strictly only valid close to  $B_{c2}$ . In order to calculate  $B_{c2}(0)$ , the WHH relation can be used where:

$$B_{c2}(0) = -0.7T_c \left. \frac{dB_{c2}}{dT} \right|_{T_c} \quad (\text{D1.5.3})$$

$B_c(0)$  can be calculated using the BCS expression

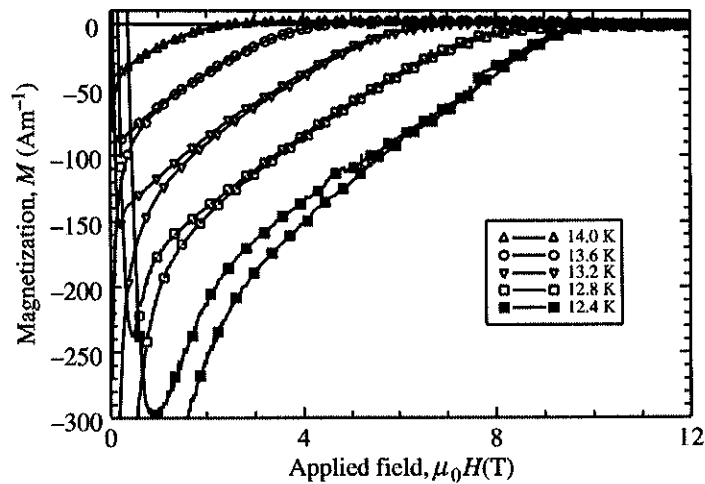


**Figure D1.5.14.** The anisotropy of the upper critical field of  $\text{SnMo}_6\text{S}_8$  at 4.2 K [99]. The angle is measured between the ternary axis and the magnetic field.

$$B_c(T) = 1.74B_c(0) \left(1 - \frac{T}{T_c}\right) \quad (\text{D1.5.4})$$

$B_{c1}(0)$  can be calculated using the Gorter–Casimir [103] two-fluid empirical relation:

$$B_{c1}(T) = B_{c1}(0) \left[1 - \left(\frac{T}{T_c}\right)^2\right] \quad (\text{D1.5.5})$$



**Figure D1.5.15.** The magnetization of bulk  $\text{PbMo}_6\text{S}_8$  as a function of field at different temperatures. From the reversible data, one can use G–L theory to calculate the G–L parameter and the upper critical field [23, 104].

**Table D1.5.3.** The values of the fundamental superconducting parameters of  $\text{PbMo}_6\text{S}_8$  derived from reversible magnetization data [104].

$T_c$	$\kappa$	$B_{c2}(0)$	$B_{c1}(0)$	$B_c(0)$	$\lambda(0)$	$\epsilon(0)$
13.7 K	130	56 T	6.4 mT	250 mT	230 nm	2.0 nm

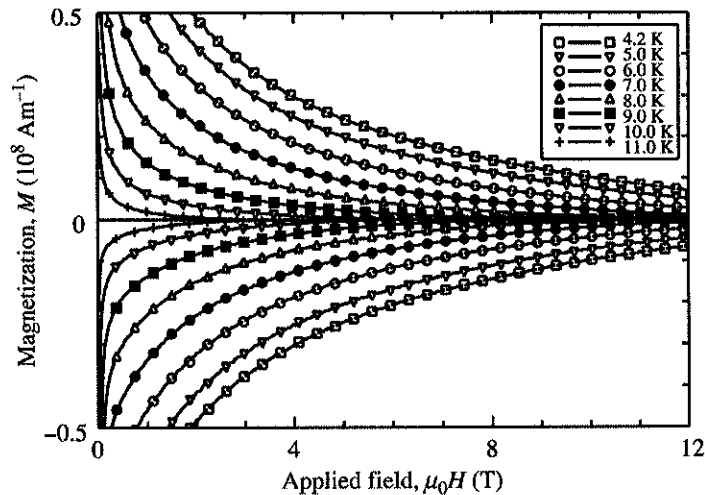
Note that this approach does mean that the G–L relations do not hold at low temperatures [104] — for example,  $B_{c2}(0) \neq \phi_0/2\pi\xi_{G-L}^2(0)$ . However, more reliable values for the critical fields are found at low temperatures using this procedure. The critical parameters for  $\text{PbMo}_6\text{S}_8$  are shown in table D1.5.3.

The magnetic penetration depth can also be measured using positive-muon spin rotation. This technique allows direct measurement of the variation in the magnetic field throughout the bulk of the sample, from which the penetration depth can be directly measured. The penetration depth has been measured in  $\text{SnMo}_6\text{S}_{8-x}\text{Se}_x$  and  $\text{PbMo}_6\text{S}_{8-x}\text{Se}_x$  [79]. There is reasonably good agreement between magnetic measurements and muon techniques.

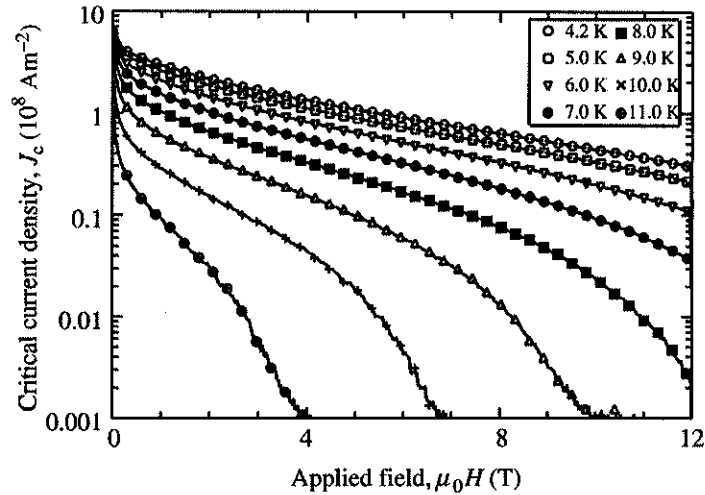
#### D1.5.8.4 Irreversibility fields

The concept of the irreversibility field ( $B_{\text{IRR}}$ ) is well documented in the literature both in the high temperature and low temperature superconductors [105, 106].  $B_{\text{IRR}}$  is the magnetic field (below  $B_{c2}$ ) at which the critical current density falls to zero. An important experimental problem is that measurements can only determine the field at which  $J_c$  drops below a minimum detection level. For practical purposes, a number of techniques are used, although the best procedure to measure  $B_{\text{IRR}}$  has not been generally agreed.

Vibrating sample magnetometry (VSM) can determine the field which delineates the hysteretic and reversible magnetic properties of a material and hence  $B_{\text{IRR}}$  directly [102]. Such measurements have been



**Figure D1.5.16.** The magnetization of bulk  $\text{PbMo}_6\text{S}_8$  as a function of field at different temperatures [23]. From the magnitude of the hysteresis, one can calculate the critical current density using Bean's model [109].



**Figure D1.5.17.** The critical current density of bulk  $\text{PbMo}_6\text{S}_8$  as a function of field at different temperatures calculated using the data from figure D1.5.14 and Bean's critical state model [23].

performed on  $\text{PbMo}_6\text{S}_8$ . In bulk  $\text{PbMo}_6\text{S}_8$ , it was found that  $B_{\text{IRR}} = 63 (1 - T/T_c)^{1.46}$  [104]. A similar power law has been observed in single crystals [105] and derived theoretically using a thermally activated flux-creep model [107]. Measurements have also been completed on  $(\text{Pb}_{1-x}\text{Gd}_x)\text{Mo}_6\text{S}_8$  that demonstrate that  $B_{\text{IRR}}$  decreased with increases in Gd content [108]. However, one should be very careful interpreting such data since the hysteresis in magnetization found in VSM measurements can only be simply related to  $J_c$  using Bean's model [109] if the field applied to the sample is uniform (cf figures D1.5.16 and D1.5.17). However as  $J_c$  drops, the hysteresis collapses to zero when the variation in applied field the sample experiences while oscillating is equal to the self field of the sample (although  $J_c$  is not zero). For example in figure D1.5.15, although the hysteresis falls dramatically at 12.8 K when the applied field reaches 0.5 T, the anomalous bump (below an extrapolation of the reversible magnetization) is associated with the inhomogeneity of the applied field and not related to  $B_{\text{IRR}}$  at all [110]. In such cases, any comparison of such field values with theoretical calculations of  $B_{\text{IRR}}$  is compromised, although it is useful to assess practical limits for high field applications.

Flux penetration measurements and transport measurements offer alternative means to measure the critical current density [111]. A type of irreversibility field can be determined by extrapolating the functional form of  $J_c$  to zero using a Kramer extrapolation [112]. In  $\text{PbMo}_6\text{S}_8$ , the irreversibility field has been improved from  $\sim 22$  T up to nearly 40 T at 4.2 K by fabricating the material using hot isostatic pressing [113].  $B_{\text{IRR}}$  values of 35.4 T at 4.2 K have been achieved in high  $J_c$  (Pb, Sn)  $\text{Mo}_6\text{S}_8$  wires [12]. Standard resistance or susceptibility measurements can also be used to determine  $B_{\text{IRR}}$  [114]. However, values obtained using different techniques can differ markedly. For example, increasing the Gd content in (Pb,Gd)  $\text{Mo}_6\text{S}_8$  increases the irreversibility field in high-fields measured using the onset of the resistive transition but shows a decrease using the onset of the susceptibility transition [111].

#### D1.5.8.5 Pinning energies

The apparent pinning energy ( $U^*$ ) of  $\text{PbMo}_6\text{S}_8$  has been measured using the decay of the magnetization in time [104].  $U^*$  varies from about 40 meV at 2 T to 15 meV at 12 T at 4.2 K. These values are about

double the equivalent values found in  $\text{YBa}_2\text{Cu}_3\text{O}_7$  and four times that of  $\text{Tl}_2\text{Ba}_2\text{Ca}_2\text{Cu}_3\text{O}_{10}$  at low fields. Activation energies derived from Arrhenius plots of resistivity give 130 meV at 9 T for  $\text{PbMo}_6\text{S}_8$  and 186 meV at 9 T and 4.2 K for  $\text{SnMo}_6\text{S}_8$  [115]. Comparisons between equivalent measurement techniques suggest that the effect of flux creep is more pronounced in  $\text{PbMo}_6\text{S}_8$  than in NbTi but less than that in high temperature superconductors.

#### D1.5.8.6 Microwave surface resistance measurements

Very few microwave measurements have been completed on Chevrel phase superconductors. For a superconducting  $\text{Cu}_2\text{Mo}_6\text{S}_8$  thin film, a surface resistance of 4.5 m $\Omega$  at 10 Ghz and 4.2 K has recently been obtained [116].

#### D1.5.8.7 Critical current versus field and temperature

The mechanism that determines the critical current density has long been a topic of theoretical and experimental research.  $J_c$  is determined both by the intrinsic fundamental superconducting properties and by the extrinsic metallurgical and microstructural factors such as the grain size of the material. Fietz and Webb [117] suggested parameterising  $J_c$  through a scaling law for the volume pinning force ( $F_p = J_c \times B$ ). The Chevrel phase materials can be described using this law where

$$F_p = J_c \times B = \alpha [B_{c2}^*(T)]^n b(1-b)^2 \quad (\text{D1.5.6})$$

where  $B_{c2}^*(T)$  is the effective upper critical field,  $\alpha$  and  $n$  are constants and  $b$  is the reduced field [ $B/B_{c2}^*(T)$ ]. The index  $n$  is typically between 2 and 3.

In low fields, the parameter  $\alpha$  increases as the grain size decreases as is also found in low temperature superconductors such as  $\text{Nb}_3\text{Sn}$  [118]. There are many different approaches to modelling the pinning including that of Kramer after whom the reduced field dependence is named but which is probably not correct in detail [112, 119]. Other pinning models have also been suggested that give the Kramer dependence and emphasize the importance of the grain boundaries. As yet, however, there is no consensus on the nature of the pinning that causes the ubiquitous Kramer dependence [120–122].

In materials optimized for high  $J_c$  in high fields, there is a much weaker correlation between  $J_c$  and grain size. For example, at 4.2 K in  $\text{SnMo}_6\text{S}_8$ ,  $J_c$  is almost independent of grain size at fields above 15 T [123]. In  $\text{PbMo}_6\text{S}_8$ ,  $J_c$  at 6 T saturates for grain sizes below 0.3  $\mu\text{m}$  [124], although it must be noted that these samples are not fully dense. Kramer found a similar saturation (or peak effect) close to  $B_{c2}$  in many low temperature superconductors [125, 126]. The value of field at which  $J_c$  extrapolates to zero [ $B_{c2}^*(T)$ ] is strongly correlated with the properties of the grain boundaries rather than either intragranular properties or the thermodynamic upper critical field [127]. Moreover in contrast to A15 commercial wires [121], during dissipation (above  $J_c$ ) flux flow is localized along narrow channels [128]. Hence for high  $J_c$  materials, the standard grain boundary description may not be appropriate in the high field (or saturation) regime. Whether this is because the efficiency of the grain boundaries falls or because a different (pinning [122, 129] or non-pinning [130]) mechanism limits  $J_c$ , remains unresolved.

The influence of neutron irradiation on  $J_c$  of Chevrel phase compounds has been investigated and some limited improvements were found [131]. The highest  $J_c$  in wires is found in the quaternary (Pb,Sn) $\text{Mo}_6\text{S}_8$  for which at 4.2 K and 14 T,  $J_c$  is  $7 \times 10^8 \text{ A m}^{-2}$  [10, 12] and at 20 T about  $2 \times 10^8 \text{ A m}^{-2}$  [132]. Although the Pb based Chevrel phase material has the highest critical field, some Sn is often included in bulk materials. This addition improves the homogeneity of the bulk [133] and the interconnectivity between the grains probably by suppressing formation of  $\text{MoS}_2$  [63]. Further improvements in the grain boundaries are still required to increase  $J_c$ . Multifilamentary wires have been produced [134, 135], as have monocoil wires with Ag [136] Ta [137] and Mo sheathing [14, 138] (cf

Chapter B3.3.5). Small three layer coils have been fabricated using  $\text{PbMo}_6\text{S}_8$  to demonstrate their potential use in magnet applications [139].

A maximum value of  $J_c$  for  $\text{PbMo}_6\text{S}_8$  has been estimated at  $10^{10} \text{ A m}^{-2}$  at 4.2 K and 20 T using a model which assumes ideal arrangement of the pinning sites [105]. Flux penetration measurements with small ac fields have found that  $J_c > 10^{10} \text{ A m}^{-2}$  at 4.2 K and 5 T at the surface of bulk  $\text{PbMo}_6\text{S}_8$  which demonstrates the potential of this material [140]. Very significant improvements in  $J_c$  have been achieved in the high temperature superconductors by fabricating textured material [141]. It remains an open question whether texturing would be useful for increasing  $J_c$  in Chevrel phase superconductors.

### D1.5.9 The magnetic Chevrel phase superconductors

Very small amounts of magnetic impurities at the parts-per-million level or paramagnetic ions at the 1 at % level are known to destroy the superconducting properties of most superconductors. Ginzburg pointed out that among the elements of the periodic table, superconductivity and magnetism seem to be mutually exclusive [142]. Early experimental work investigating superconductors with magnetic impurities was compromised by uncertainty over whether or not the superconductivity and the ferromagnetism coexisted in the same region of the sample. In Chevrel phase rare-earth materials, however, superconductivity and magnetism coexist within the same unit cell.

The structure of magnetic Chevrel phase superconductors with the rare-earth element centrally located in the unit cell gives a relatively large distance, and hence weak overlap, between the 4d-electrons of the Mo and the 4f-electrons of the rare-earth element. In contrast, the 3d-elements are located close to the  $\text{Mo}_6\text{S}_8$  cluster and the superconductivity is destroyed. At present there are no single crystals of heavy rare-earth Chevrel phase selenides available (cf section D1.5.5).

Susceptibility measurements show that the ternary magnetic superconductors have an effective Bohr magneton number that is close to the theoretical values for isolated ions at temperature above about 50 K [143, 144]. At lower temperatures, deviations from the Curie–Weiss law occur because of crystal field and magnetic correlation effects. There are many similarities in the superconducting and magnetic properties of the rare-earth Chevrel phase materials and the strongly magnetic nickel–boron-carbide materials [145] and the Rare-earth-rhodium-borides. The properties of  $\text{GdMo}_6\text{S}_8$  are shown in figure D1.5.18 [4, 6]. The re-entrant resistance is correlated with the antiferromagnetic ordering which occurs at 0.82 K as shown by the heat capacity measurements and neutron scattering measurements [4, 146, 147]. The nature of the magnetic ordering is dependent on the particular rare-earth element in the compound and can be antiferromagnetic, ferromagnetic or oscillatory. For example in  $\text{HoMo}_6\text{S}_8$ , tunnelling spectrometry suggests that superconductivity coexists with ferromagnetism [148]. In high fields, so called re-entrant superconductivity can occur in some systems [149]. It has been suggested that in high fields, the applied external field compensates for the negative exchange interaction between the rare-earth ion and the conduction electrons so that the material becomes superconducting [150] in a limited part of  $B$ – $T$  phase space. The degree to which ordering of the ions is exchange driven [151] and driven by dipolar interactions [149] is still open to discussion.

The range of phenomena of the rare-earth Chevrel phase superconductors continues to fascinate the scientific community. The complexity arises because in order to understand the properties of these materials we must understand how magnetism and superconductivity operate at the atomic level.

### D1.5.10 Concluding comments

Since their discovery in 1971, Chevrel phase materials have been of interest to the whole superconductivity community from engineers, who want to make high field magnets, to physicists, who want

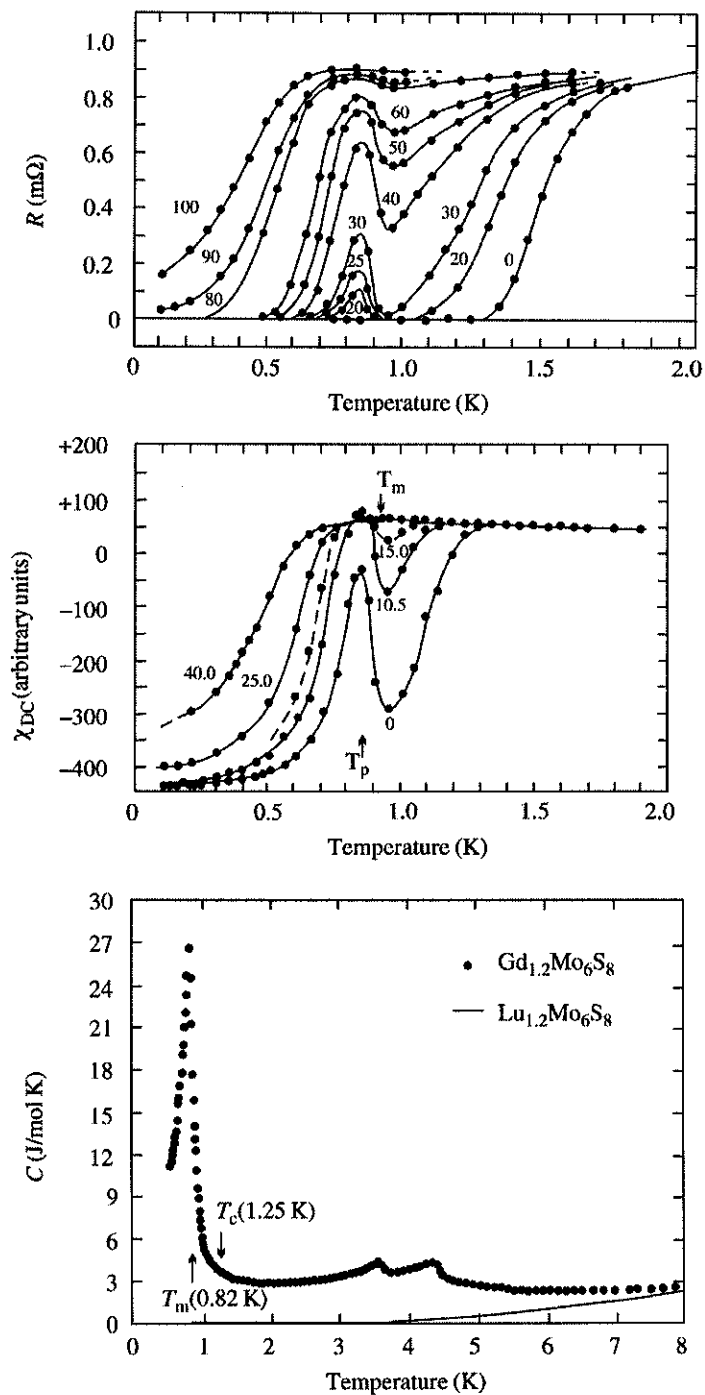


Figure D1.5.18. Selected properties of the magnetic rare-earth compound  $GdMo_6S_8$  [6].

to understand the microscopic mechanism that causes superconductivity, magnetism and coexistence of the two. Interest in these materials inevitably waned with the discovery of the high temperature superconductors in 1986 by Bednorz and Muller. Massive interest in HTS materials was driven by the possibility of discovering a new mechanism producing superconductivity and the potential for new applications operating at liquid nitrogen temperatures. Related driving forces are now increasing the research activity into Chevrel phase superconductors. Fundamental interest arises because these materials may offer a model (almost) cubic system in which to address non-BCS superconductivity without the strong anisotropy or layering present in the HTS materials. Furthermore, Chevrel phase superconductors have interesting fundamental properties that are intermediate between the HTS and LTS materials. Technological interest arises because of the high values of  $B_{c2}$  in these materials and recent improvements in cryocooler technology which facilitates operating very high field magnets at  $\sim 4$  K [152]. Indeed if  $J_c$  in the wires of these materials can be improved by a factor of  $\sim 4$ , we can expect to use them in the next generation of high field magnet systems operating in fields significantly above 25 T.

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**Further Reading**

Maple M B and Fischer Ø 1982 *Superconductivity in Ternary Compounds Vol. I and II – Structural, Electronic and Lattice Properties* Berlin: Springer

These two volume texts were written by many of the individual researchers involved in the intensive research of the seventies. The texts include an excellent compendium of many of the material properties for Chevrel phase superconductors — to avoid extending our reference list considerably we have quoted these texts as a source rather than citing the several hundred papers quoted therein.

Evetts J 1992 *Concise Encyclopaedia of Magnetic and Superconducting Materials* Oxford, UK: Pergamon

Series of compact, introductory articles predominantly on the science and technology of magnetic and superconducting materials.

Pena O Sergeant M 1989 Rare earth based Chevrel phases  $REMo_6X_8$ : crystal growth, physical and superconducting properties *Prog. Solid State Chem.* **19** 165–281

A very interesting comprehensive review of single crystal growth and properties. Great detail is provided on the chemistry and materials science of rare earth materials.

Fischer Ø 1978 Chevrel phases: superconducting and normal state properties *Appl. Phys.* **16** 1–28

Excellent older review of the properties of Chevrel phase superconductors.